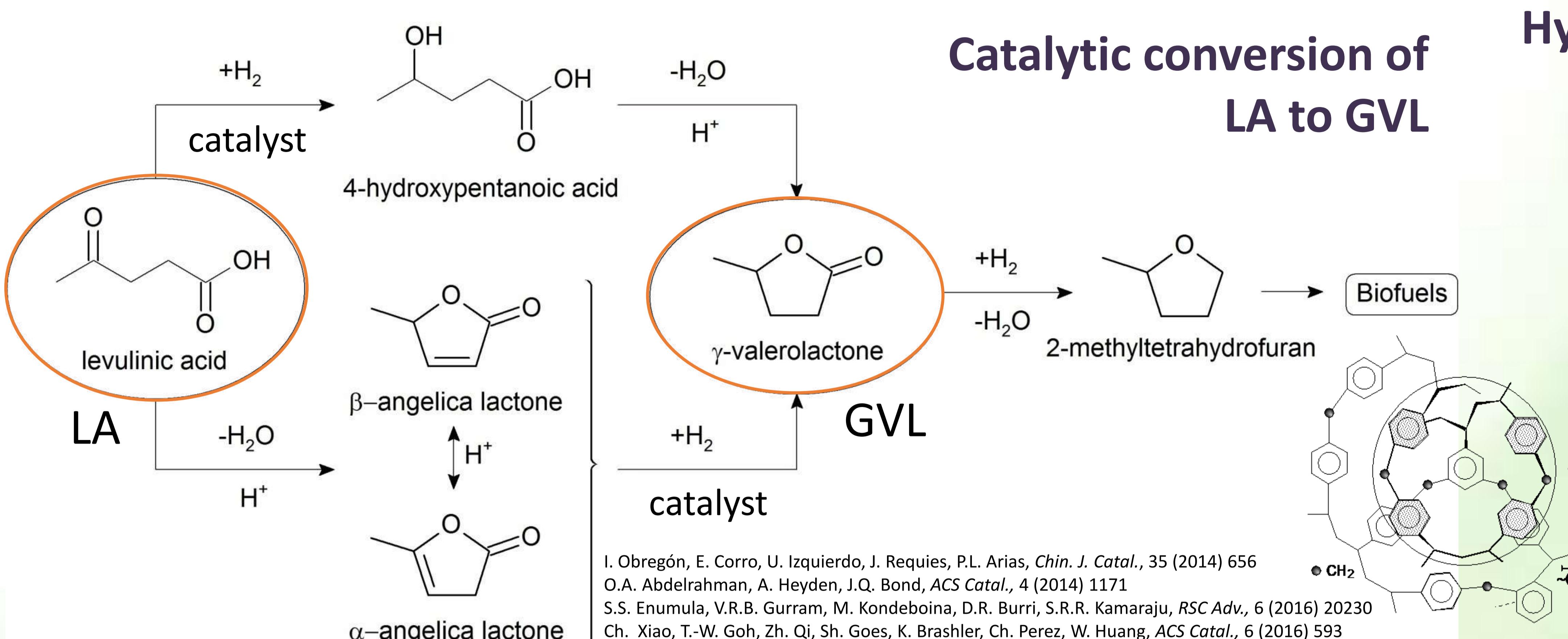


MONO- AND BIMETALLIC CATALYSTS BASED ON HYPER-CROSSLINKED POLYSTYRENE FOR HYDROGENATION OF BIOMASS-DERIVED LEVULINIC ACID

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Hypercrosslinked polystyrene (HPS) as catalytic support

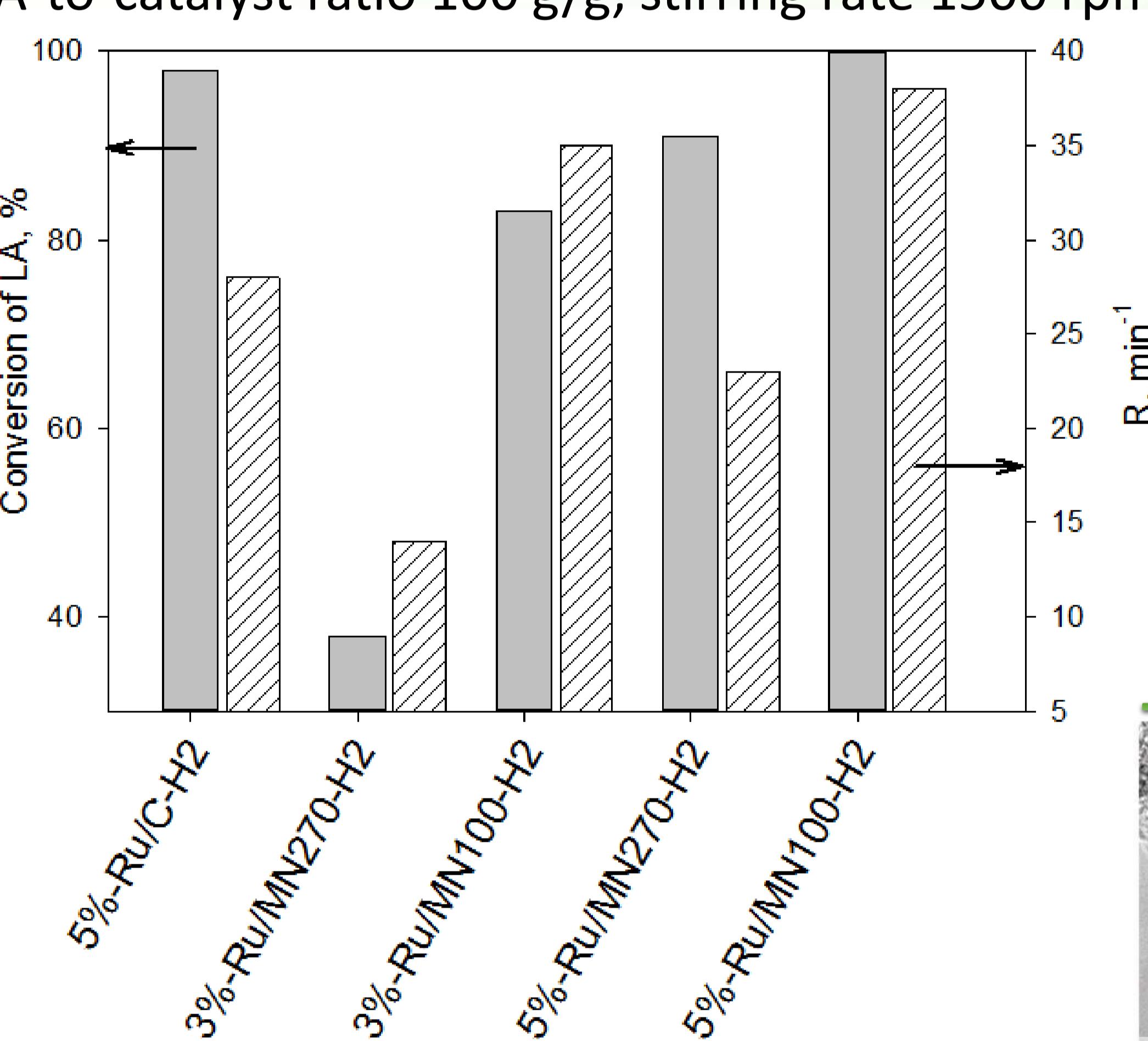
Synthesized Ru/HPS catalysts:

- 3%-Ru/MN100 All the samples were activated in H_2 flow ($300^\circ C$, duration 2 h)
- 5%-Ru/MN100
- 3%-Ru/MN270
- 5%-Ru/MN270

The following types of HPS produced by Purolite Ltd. (UK) were used:

- MN100 (bearing tertiary amino-groups)
- MN270 (non-functionalized)

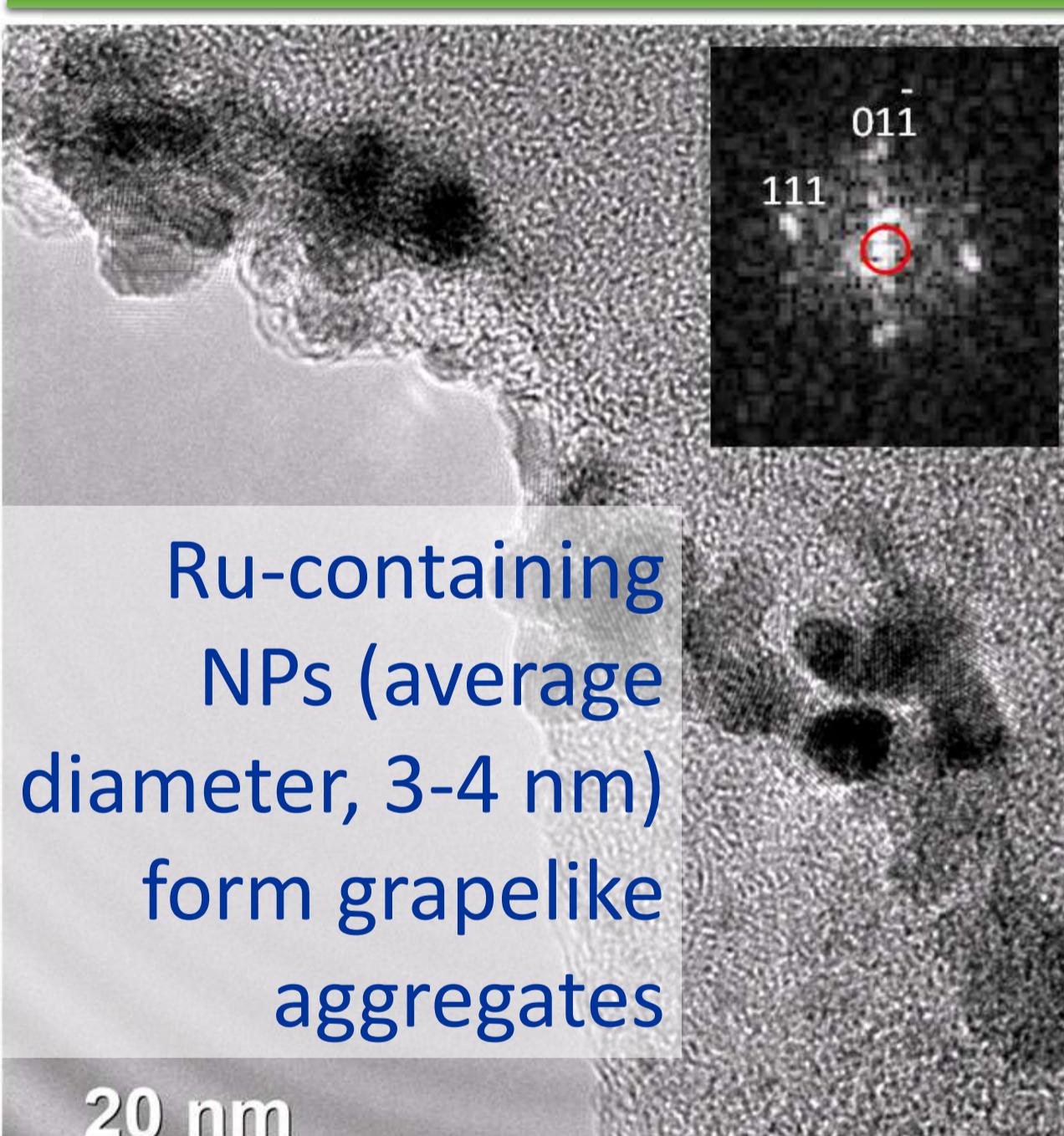
temperature $100^\circ C$, hydrogen partial pressure 2 MPa, LA-to-catalyst ratio 100 g/g, stirring rate 1500 rpm



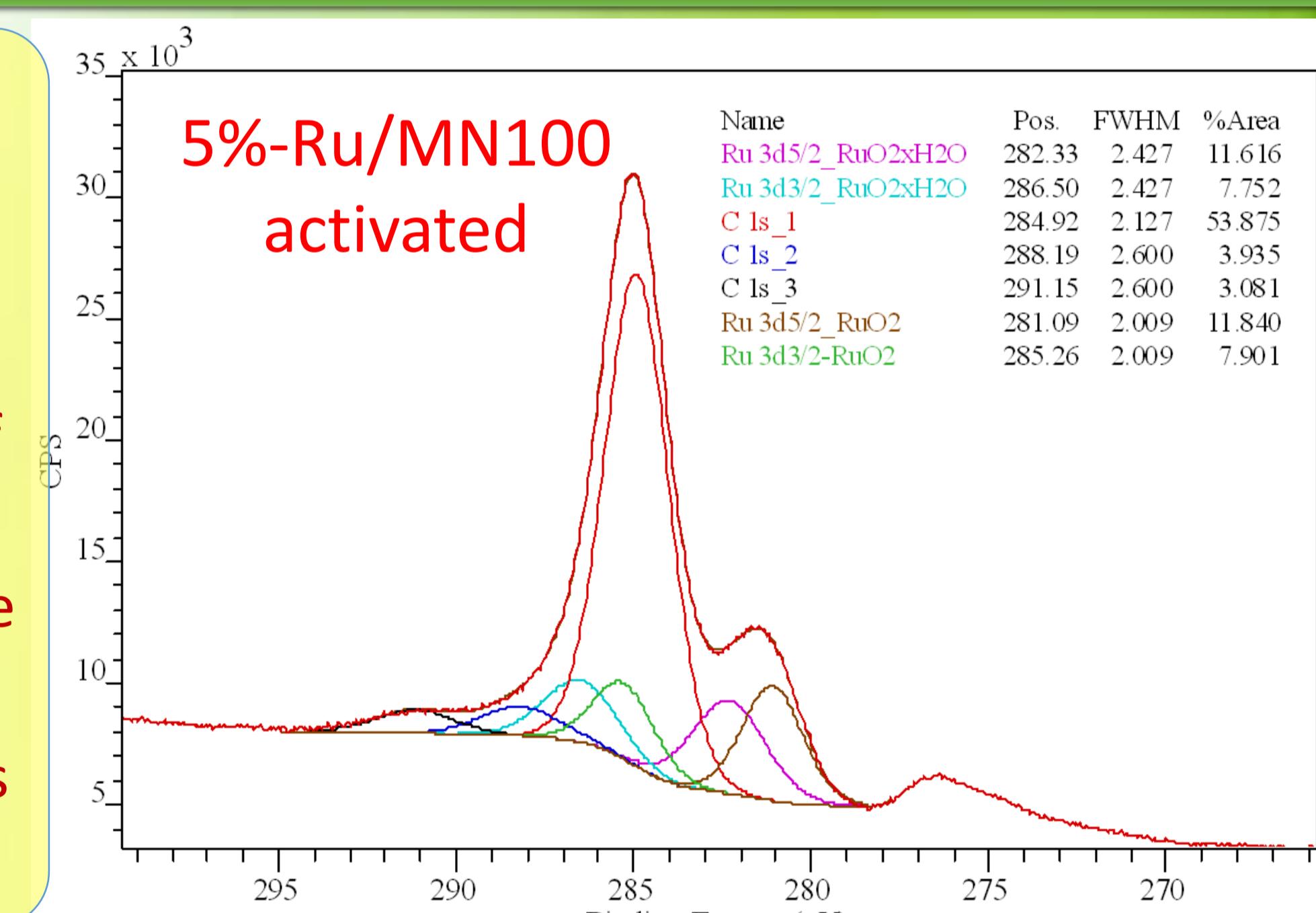
Catalyst (activated with H_2)	X_{LA} , % (after 100 min of reaction time)	R , min^{-1}
5%-Ru/C (Acros Organics)	98	28
3%-Ru/MN270	38	14
3%-Ru/MN100	83	35
5%-Ru/MN270	91	23
5%-Ru/MN100	~100	38

Selectivity was ~100% in all the experiments

$$R (\text{mol}_{LA} \cdot \text{mol}_{Ru}^{-1} \cdot \text{min}^{-1}) = (N_{LA,X2} - N_{LA,X1}) \times N_{Ru}^{-1} \times (\tau_2 - \tau_1)^{-1}$$

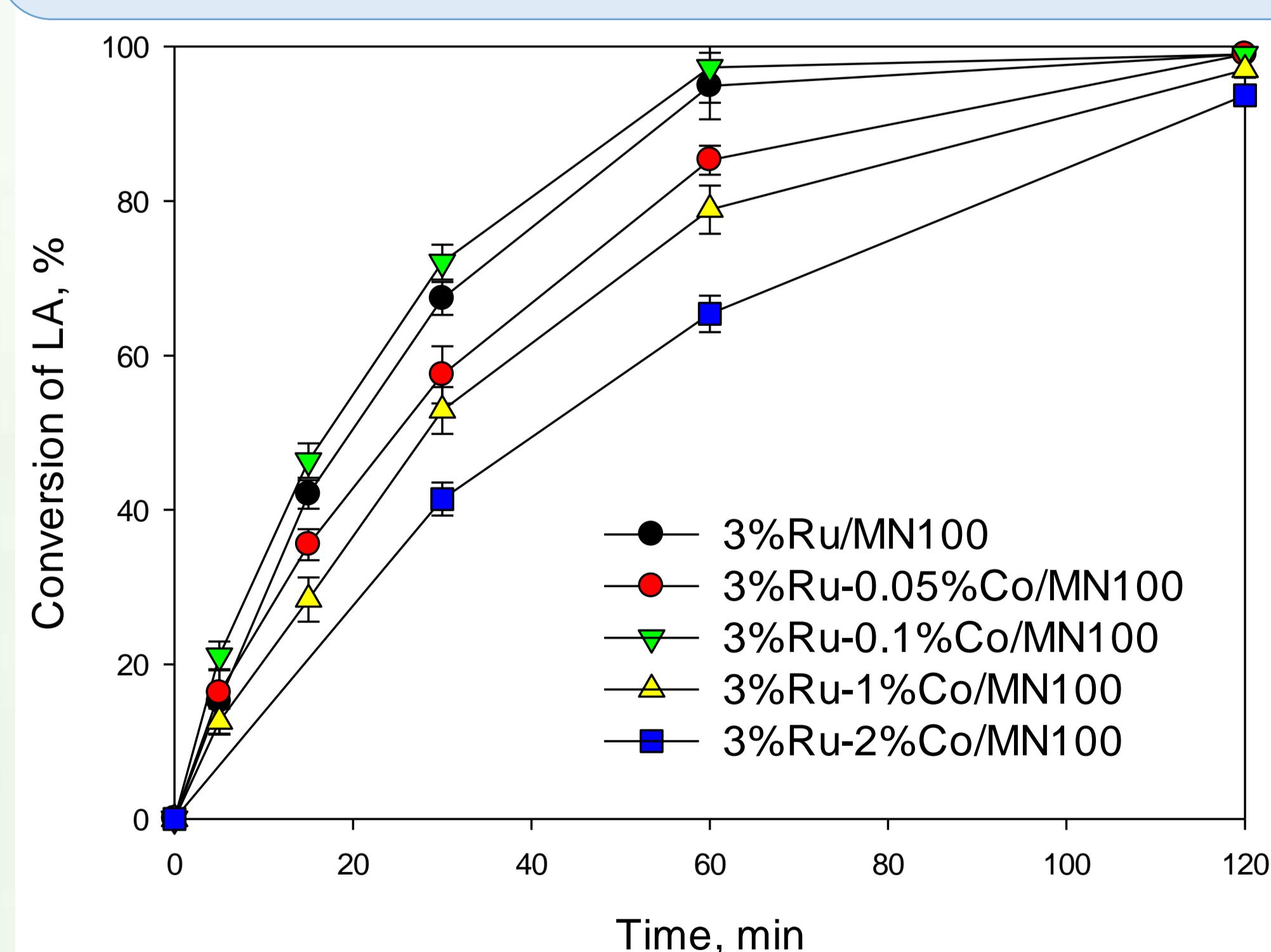


Ru(0) is absent on the surface of activated 5%-Ru/MN100, thus the activity of 5%-Ru/MN100 can be attributed to the formation of small RuO₂ nanoparticles (NPs)

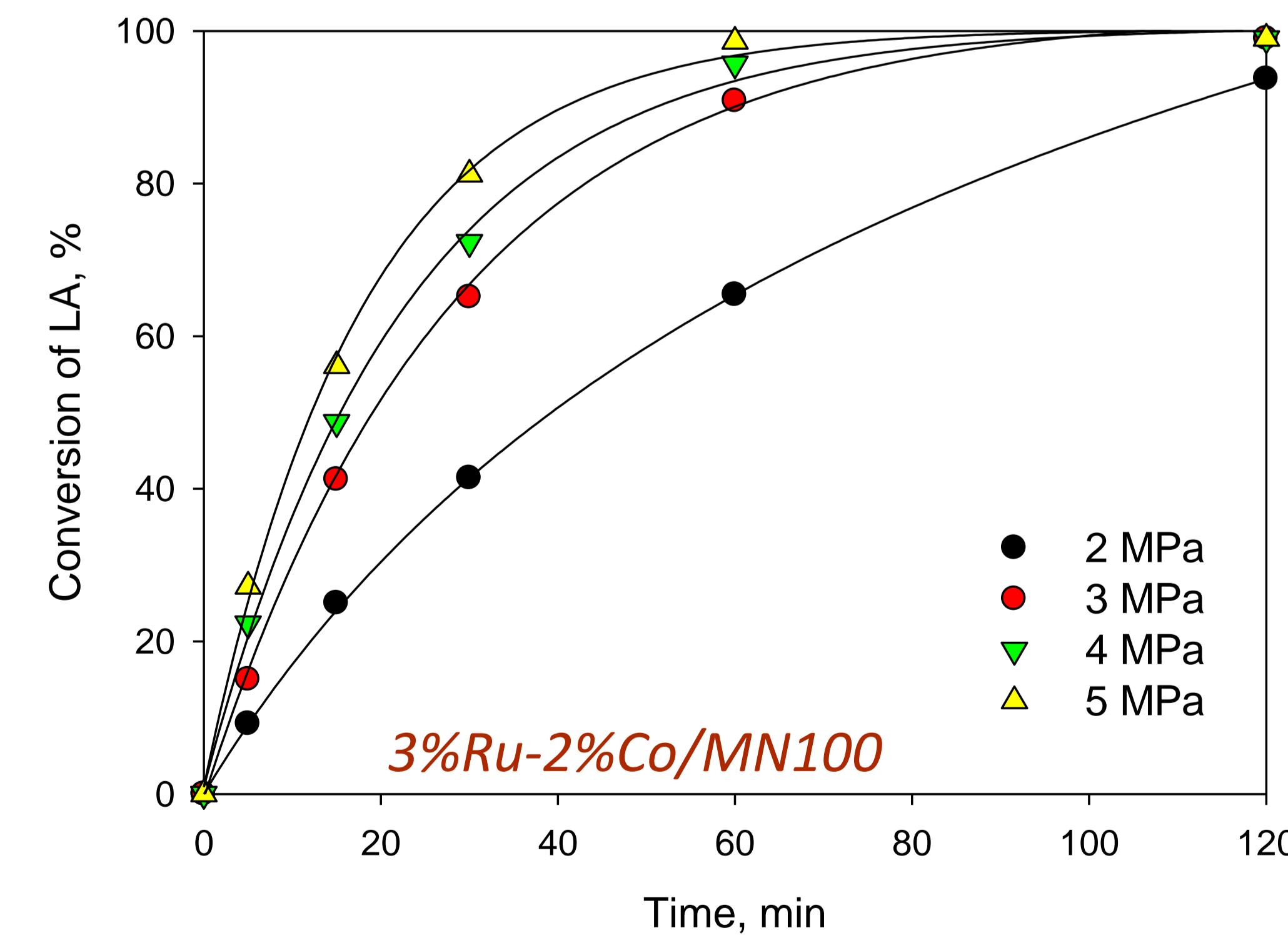
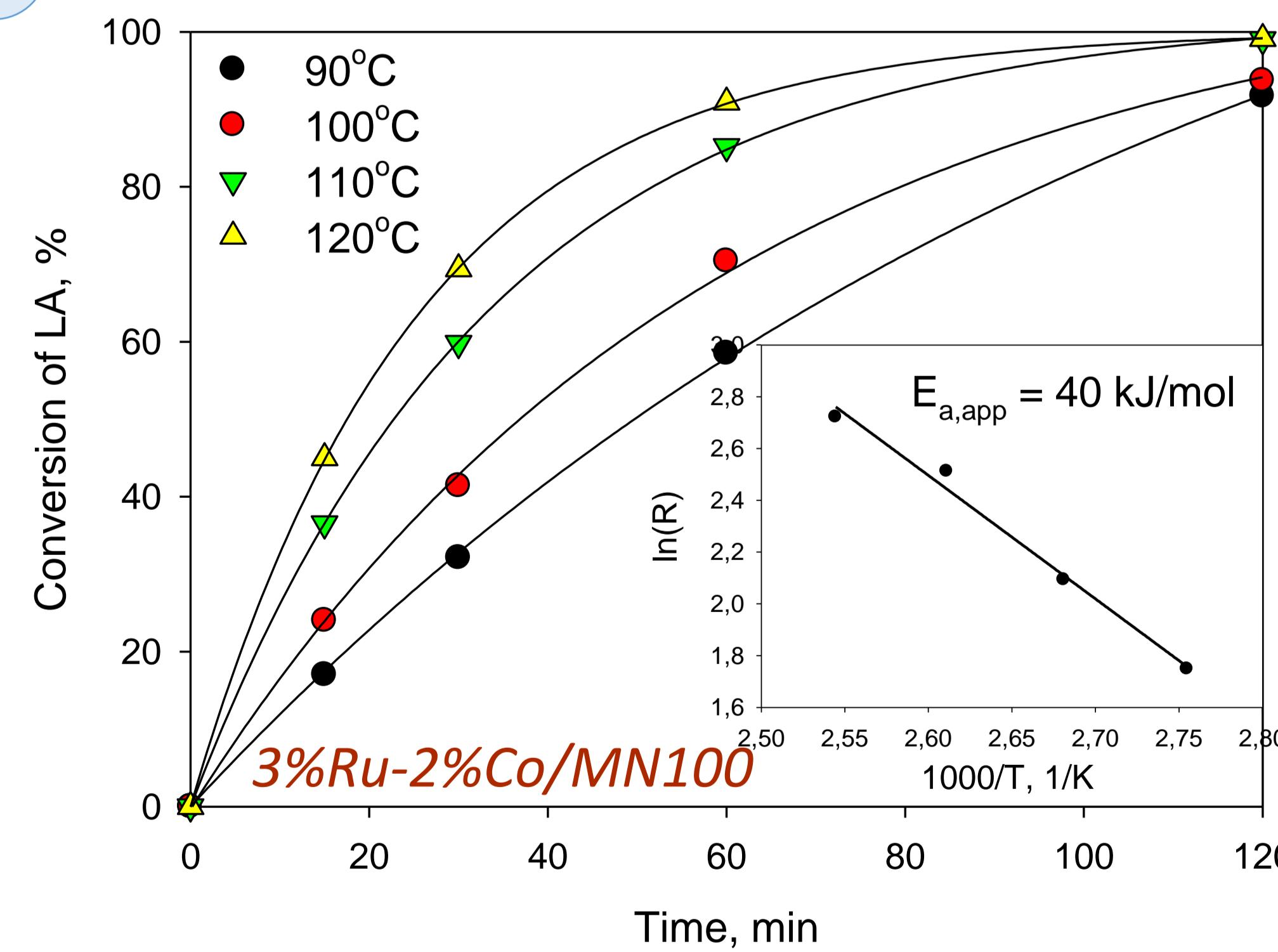


Synthesized Ru-Co/HPS catalysts:

- 3%Ru-0.05%Co/MN100 All the samples were activated in H_2 flow ($300^\circ C$, duration 2 h)
- 3%Ru-0.1%Co/MN100
- 3%Ru-1%Co/MN100
- 3%Ru-2%Co/MN100



100°C, H_2 partial pressure 2 MPa, LA-to-catalyst ratio 50 g/g



Introduction of Co in the catalyst composition results in redistribution of nanoparticles of catalytically active phase (RuO_2) inside the polymeric matrix of HPS. No products of chemical interaction of Co- and Ru-containing species were found on the catalyst surface

M.E. Grigorev et al., *Catalysis Today*, 2020, in press

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